

9#-Lanostane-Type Triterpene Lactones from the Stem Bark of *Abies veitchii*

R. Tanaka, and S. Matsunaga

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9 β -LANOSTANE-TYPE TRITERPENE LACTONES FROM THE STEM BARK OF *ABIES VEITCHII*

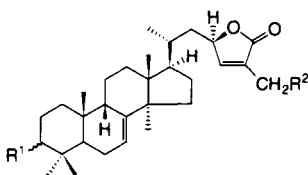
R. TANAKA and S. MATSUNAGA*

Osaka University of Pharmaceutical Sciences, 2-10-65 Kawai, Matsubara, Osaka 580, Japan

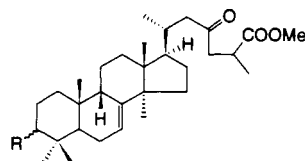
ABSTRACT.—Two new tetracyclic triterpene lactones were isolated, together with two known compounds, 3-oxo-9 β -lanosta-7,24-dien-26,23*R*-olide [**2**] and 3 β -hydroxy-9 β -lanosta-7,24-dien-26,23*R*-olide [**3**], from the stem bark of *Abies veitchii*. The structures of the new compounds were established as 3 α -hydroxy-9 β -lanosta-7,24-dien-26,23*R*-olide [**6**] and 3 α -methoxylanosta-7,9(11),24-trien-26,23*R*-olide [**7**], respectively, on the basis of chemical and spectral evidence.

Abies veitchii Lindl. (Pinaceae) (Japanese name: Shirabiso) is a tall evergreen tree growing deep in the mountains from central to northern Japan (1). Our previous studies on the chemical constituents of *Abies* species have revealed a tetracyclic triterpene lactone, abieslactone [**1**] (3 α -methoxy-9 β -lanosta-7,24-dien-26,23*R*-olide) (2,3), from the bark and leaves of *Abies mariesii* Mast., *Abies amabilis* (Dougl.) Forbes, and *Abies procera* Rehd., a mixture of five *n*-alkyl ferulates bearing alkyl moieties from C-22 to C-26, a mixture of campesterol and β -sitosterol, 3-oxo-9 β -lanosta-7,24-dien-26,23*R*-olide [**2**] (4), 3 β -hydroxy-9 β -lanosta-7,24-dien-26,23*R*-olide [**3**], and 27-hydroxy-3-oxo-9 β -lanosta-7,24-dien-26,23*R*-olide [**4**] (5) from the bark of *Abies firma* Sieb. et Zucc., and abieslactone [**1**] and veitchiolide [**5**] [7 β -hydroxy-3 α -methoxylanosta-9(11),24-dien-26,23*R*-olide] from the bark of *A. veitchii* (6). The fact that the bark of *A. firma* contained substantial **2** and lacked abieslactone [**1**], which is the most abundant triterpenoid in the bark of both *A. mariesii* and *A. veitchii*, drew our chemotaxonomic interest in this species. During the course of our work, five groups of investigators have reported the isolation and the structure elucidation of several highly oxygenated lanostanes, 9 β -lanostanes, cycloartanes, and some of their analogues from the bark and needles of *Abies alba* Mill. (7,8), the bark of *A. grandis* (9), the seeds of both *A. mariesii* and *A. firma* (10,11), and the bark and needles of *Abies sibirica* (12,13).

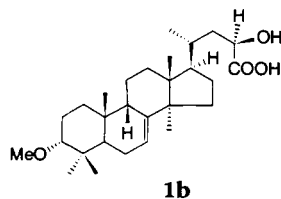
Recently, we reported that methyl esters **1a** and **2a** of two γ -keto acids and a trisnor- α -hydroxy acid **1b**, derived from compounds **1** or **2** (2), strongly inhibited both (a) incorporation of 32 P into phospholipids of HeLa cells cultured in the medium containing 12-*O*-tetradecanoylphorbol-13-acetate (TPA) and inorganic 32 P and (b) the promoting action of TPA on skin tumor formation in mice initiated with 7,12-dimethyl-



- | | |
|------------------|--|
| 1 | R ¹ = α -OMe, R ² = H |
| 2 | R ¹ = $>$ O, R ² = H |
| 3 | R ¹ = β -OH, R ² = H |
| 4 | R ¹ = $>$ O, R ² = OH |
| 6 | R ¹ = α -OH, R ² = H |
| 6-acetate | R ¹ = α -OAc, R ² = H |



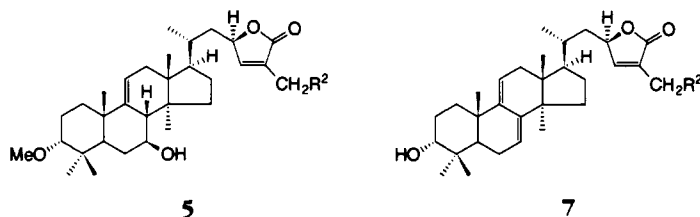
- | | |
|-----------|-------------------|
| 1a | R = α -OMe |
| 2a | R = $>$ O, |



1b

benz[*a*]anthracene (14), although compounds **1** and **2** themselves showed no remarkable activity in the same *in vitro* assay.

In the search for stronger and more effective antitumor-promoting agents from plant sources, we re-examined the neutral Et₂O extract of the stem bark of *A. veitchii* and isolated two unknown triterpenoid lactones **6** and **7**, together with two known triterpene constituents **2** and **3**. This paper deals with the structure elucidation of compounds **6** and **7**.



RESULTS AND DISCUSSION

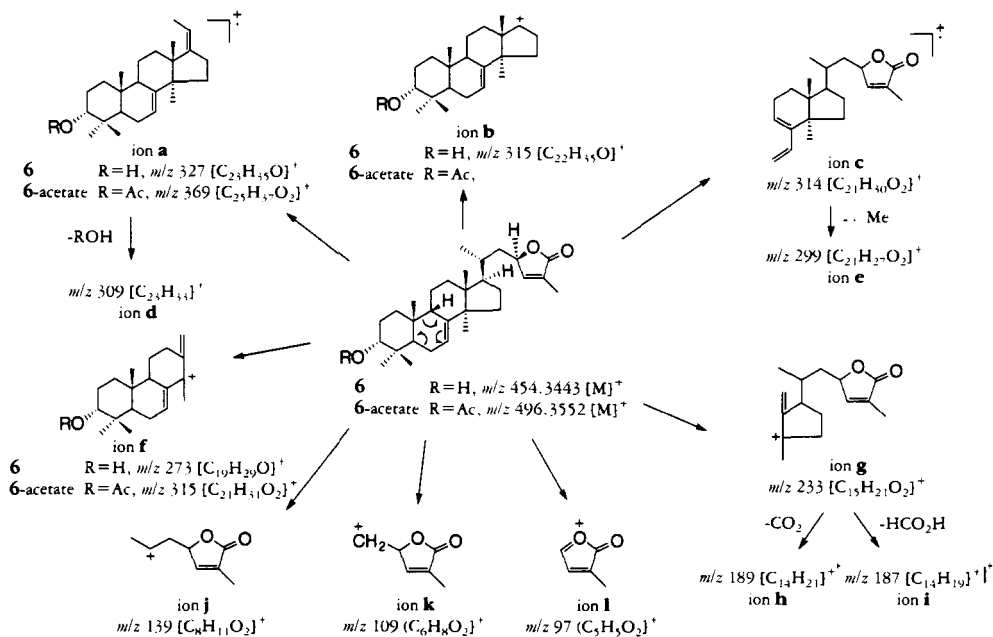
The known triterpenoids were identified by direct comparison with samples of 3-oxo-9 β -lanosta-7,24-dien-26,23*R*-olide [**2**] and 3 β -hydroxy-9 β -lanosta-7,24-dien-26,23*R*-olide [**3**], respectively, isolated from the bark of *A. firma* (4,5).

Compound **6**, one of two unknown compounds, gave the molecular formula C₃₀H₄₆O₃ ($[M]^+$ at m/z 454.3443) and showed positive color on the Liebermann-Burchard test. Its ir, ¹H-nmr (Table 1), and ¹³C-nmr spectra (Table 2) exhibited the presence of five tertiary methyl groups, a secondary methyl group, a secondary carbinolic methine group [δ 3.43 (1H, narrow diffused t, $W/2 = 6.8$ Hz) and 76.32 (H-C-OH)], a trisubstituted olefin bond [δ 5.55 (1H, H-7), 121.64 (=CH-) and 148.60 (=C<)], along with signals due to the presence of a 4-substituted 2-methyl-2-butenolide ring (closely similar to that in both compounds **2** and **3**) at δ 1.91 (3H, t, Me-27), 4.98 (1H, ddd, H-23), 7.00 (1H, H-24), 129.46 (=C<) and 149.72 (=CH-). In the DEPT and eims, compound **6** showed the same carbon composition and fragment ion peaks (Scheme 1) closely similar to those of compound **3** (5). However, ¹H-nmr chemical shift values for H-3, Me-19, Me-28, and Me-29 in compound **6** showed significant discrepancies of $\Delta\delta + 0.17$, $+ 0.08$, $+ 0.12$, and $- 0.09$ ppm, respectively, in comparison with those of compound **3**, although all the other ¹H signals in both compounds agreed within ± 0.01 ppm. Furthermore, ¹³C-nmr signals of C-1, C-2, C-3, C-4, C-5, C-19, C-28, and C-29 around the A ring in compound **6** showed chemical shift values considerably different from those of compound **3**, whereas all the other signals of both compounds were in good agreement. Acetylation of compound **6** yielded **6**-acetate. In the ¹H-nmr spectrum, compounds **6** and **6**-acetate exhibited signals attributable to 3-equatorial methine protons geminal to the hydroxyl group and acetoxy group as narrow diffused triplets ($W/2 = 6.8$ Hz) at δ 3.43 and 4.65, respectively, while compound **3** showed the corresponding 3-axial methine proton signal as a double doublet ($J = 10.2$ and 5.5 Hz) at δ 3.20 (5). All these data supported the theory that compound **6** must be the 3 α isomer of **3**. The 23*R* configuration of the lactone side chain in compound **6** was estimated from cd measurements in which it gave a negative Cotton effect curve similar to those of compounds **2** and **3** (4,5). Definite proof for this assumption was obtained by the following experiment. Chromium trioxide oxidation of **6** in pyridine afforded the keto-lactone identical in all respects with compound **2**. Accord-

TABLE 1. ¹H-nmr Chemical Shifts of Compounds 2, 3, 5, 6, 6-acetate, and 7 in CDCl₃.

Proton	Compound						
	2	3	5 ^a	6	6-acetate	7	
Me-18	0.81	0.93	0.69	0.943	0.95	0.59	
Me-19	1.00	0.93	1.08	1.01	1.02	1.00	
Me-21	1.01	1.00	1.03	1.00	1.00	1.02	
	(d, 6.5)	(d, 6.5)	(d, 6.5)	(d, 6.5)	(d, 6.5)	(d, 6.5)	
Me-27	1.92	1.91	1.92	1.92	1.91	1.92	
	(t, 1.7)	(t, 1.7)	(t, 1.7)	(t, 1.7)	(t, 1.7)	(t, 1.7)	
Me-28	1.10	0.87	0.94	0.99	0.88	0.948	
Me-29	1.09	1.02	0.89	0.925	0.98	0.934	
Me-30	1.03	1.02	0.83	1.03	1.02	0.87	
H-2 α , β	2.49	—	—	—	—	—	
	(dd, 8.5, 6.5)						
H-3	—	3.20	2.84	3.43	4.65	2.84	
		(dd, 10.0, 5.5)	(t, 2.4)	(dif. t, W/2 6.8)	(dif. t, W/2 6.8)	(dif. t, W/2 6.8)	
H-7	5.64	5.56	3.71	5.55	5.55	5.31	
	(dt, 6.5, 3.0)	(dt, 6.5, 3.0)	(td, 11.5, 5.3)	(dt, 6.5, 3.0)	(dt, 6.5, 3.0)	(br. d, 6.1)	
H-11	—	—	5.31	—	—	5.44	
			(br. d, 6.3)			(t, 4.5)	
H-23	4.98	4.97	4.99	4.98	4.97	4.98	
	(ddd, 9.4, 4.1, 1.7)	(ddd, 9.4, 4.1, 1.7)	(ddd, 9.4, 4.1, 1.7)	(ddd, 9.4, 4.1, 1.7)	(ddd, 9.4, 4.1, 1.7)	(ddd, 9.5, 4.1, 1.7)	
H-24	7.01	6.99	7.00	7.00	7.00	7.01	
	(quint, 1.7)	(quint, 1.7)	(quint, 1.7)	(quint, 1.7)	(quint, 1.7)	(quint, 1.7)	
3 α -OMe	—	—	3.31	—	—	3.29	
3 α -OAc	—	—	—	—	2.05	—	

^aData for this compound are taken from Tanaka and Matsunaga (6).

SCHEME 1. Mass spectral fragmentation of compounds **6** and **6-acetate**.

ingly, the structure of compound **6** was proved to be 3 α -hydroxy-9 β -lanosta-7,24-dien-26,23*R*-olide.

Compound **7**, the second new compound, gave the molecular formula C₃₁H₄₆O₃ ([M]⁺ at m/z 466.3451) and also was positive in the Liebermann-Burchard test. Its uv and ir spectra showed the presence of an α,β -unsaturated γ -lactone ring [λ max 219 nm; ν max 1758 (shoulder) and 1740 cm⁻¹] and a heteroannular diene chromophore (λ max 236, 243.5 and 251.5 nm; ν max 1658, 897 and 815 cm⁻¹) in the molecule. In the ¹H- and ¹³C-nmr spectra, it exhibited signals due to five quaternary methyl groups, a secondary methyl group, a vinylic methyl group deshielded with the lactone carbonyl, a secondary methoxy group attached to the C-3 α position [δ 2.84 (1H, narrow diffused t, H-C-OMe), 3.29 (3H, s, OMe) and 56.87 (C-3)], a -H₂C-HC=C-C=CH-CH₂- moiety [δ 5.31 (1H, H-7), 5.44 (1H, H-11), 115.27 and 120.49 (each =CH-), and 142.28 and 146.23 (each =C<)], a trisubstituted olefin bond [δ 7.01 (1H, H-24), 129.40 (=C<), and 149.78 (=CH-)] lying between the lactone carbonyl [δ 174.42 (C-26)] and the methine proton attached to the etheral oxygen in the γ -lactone ring [δ 4.98 (1H, H-23)] similar to those of the 4-substituted 2-methyl-2-butenolide ring in compound **6**. Detailed analysis of 2D COSY and eims for compound **7** furnished appropriate skeletal information. In the 2D long range ¹H-¹³C nmr spectrum, compound **7** showed cross correlation for signals of Me-18 (with C-12, C-13, C-14, and C-17), Me-19 (with C-1, C-5, C-9, and C-10), Me-21 (with C-17, C-20, and C-22), Me-27 (with C-24, C-25, and C-26), Me-28 (with C-3, C-4, C-5, and C-29), Me-29 (with C-3, C-4, C-5, and C-28) and Me-30 (with C-8, C-13, C-14, and C-15), as well as H-7 (with C-6 and C-8) and H-11 (with C-9 and C-12). The eims of compound **7** exhibited characteristic fragment ion peaks considered to arise from the cleavage of the lanosta-7,9(11)-diene skeleton at m/z 391 [M - Me - MeOH - CO]⁺, 339 [ion **m**], 337 [ion **n**], 327 [ion **o**], 325 [ion **p**], 295 [ion **q**], 285 [ion **r**], and 253 [ion **s**], together with peaks due to cleavage of the side chain moiety containing a 2-methyl-2-butenolide ring

TABLE 2. ^{13}C -nmr Chemical Shifts of Compounds **2**, **3**, **5**, **6**, **6**-acetate, and **7** in CDCl_3 (TMS = 0).^a

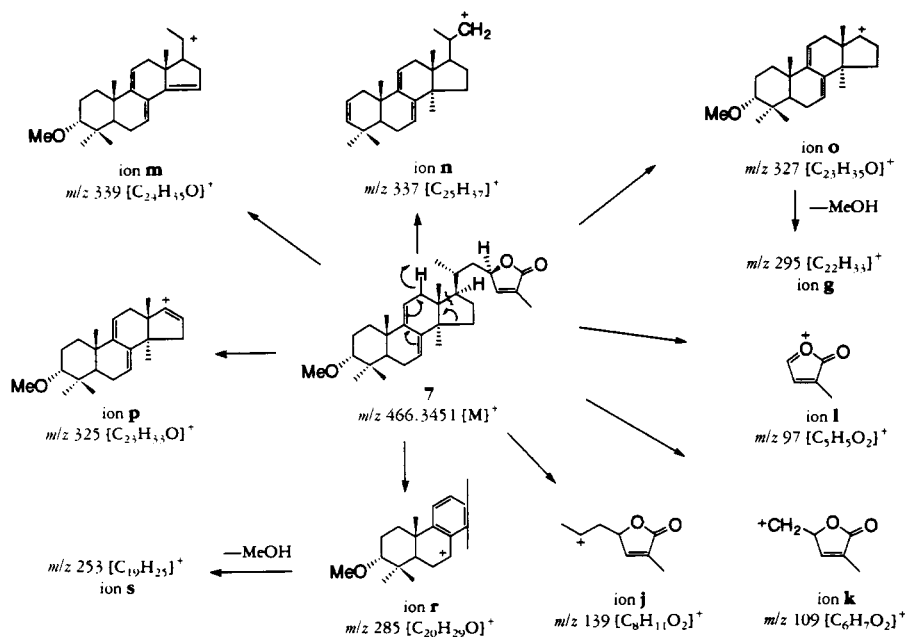
Carbon	Compound					
	2	3	5^b	6	6 -acetate	7
C-1	34.20	35.52	31.00	29.82	30.54	30.06
C-2	34.28	27.94	20.43	25.72	23.23	20.35
C-3	218.92	79.33	85.51	76.32	78.45	85.67
C-4	47.01	38.83	37.97	37.40	36.50	37.78
C-5	52.35	48.62	42.55	43.53	43.33	43.72
C-6	23.01	23.07	31.30	23.11	23.05	22.99
C-7	121.64	121.70	72.31	121.64	121.51	120.49
C-8	148.48	148.59	50.39	148.60	148.53	142.28
C-9	45.67	48.33	146.43	48.53	48.42	146.23
C-10	35.80	35.90	39.10	35.69	35.58	37.26
C-11	20.88	22.93	116.56	22.89	22.85	115.27
C-12	34.39	35.26	36.83	35.35	35.27	37.65
C-13	44.15	43.74	45.24	43.73	43.72	43.95
C-14	51.93	52.76	46.62	52.88	52.86	50.47
C-15	33.01	33.32	36.61	33.28	33.24	31.44
C-16	28.21	28.55	28.58	28.58	28.57	27.88
C-17	53.47	53.96	50.55	53.98	53.98	51.39
C-18	22.51	23.66	14.36	23.77	23.78	15.73
C-19	23.12	24.49	21.89	24.37	24.31	22.71
C-20	33.49	33.47	33.33	33.47	33.47	33.55
C-21	18.13	18.41	18.41	18.39	18.41	18.44
C-22	40.46	40.46	40.73	40.46	40.41	40.60
C-23	78.96	79.01	78.91	79.02	79.02	78.97
C-24	149.68	149.66	149.69	149.72	149.72	149.78
C-25	129.48	129.51	129.52	129.46	129.43	129.40
C-26	174.44	174.43	174.39	174.44	174.42	174.44
C-27	10.65	10.65	10.65	10.63	10.63	10.63
C-28	21.30	16.36	28.22	28.66	28.26	28.27
C-29	28.00	28.89	22.73	23.41	23.05	23.15
C-30	27.41	30.45	18.27	30.72	30.80	25.71
OMe	—	—	57.01	—	—	56.87
CO ₂ Me	—	—	—	—	21.30	—
CO ₂ Me	—	—	—	—	170.67	—

^aAssignments were made by 2D ^1H - ^1H COSY, 2D ^1H - ^{13}C COSY, and 2D long-range ^1H - ^{13}C COSY experiments.

^bData for this compound are from Tanaka and Matsunaga (6).

analogous to those of compounds **2** and **3** at m/z 139 [ion **j**], 109 [ion **k**], and 97 [ion **l**] (Scheme 2). On the other hand, the cd curve of compound **7** gave a negative Cotton effect (see Experimental) similar to those of compounds **1**–**6** (1, 3–5), indicative of the same side chain containing the 23*R* configuration of the lactone ring with these compounds. All these data indicated that compound **7** must be 3 α -methoxylanosta-7,9(11),24-trien-26,23*R*-olide. Conclusive evidence for this structure was obtained from the following experiment. Oxidation of abieslactone [**1**] with selenium dioxide in HOAc furnished the 7,9(11),24-trien-26,23*R*-olide, which was identical in all respects with compound **7**.

To the best of our knowledge, compounds **6** and **7** have not yet been described in the literature. Further investigation on the anti-tumor-promoting activity for **6**, **7**, and their derivatives is now in progress, and results will be reported elsewhere.



SCHEME 2. Mass spectral fragmentation of compound 7.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Melting points were measured on a Yanagimoto micro-melting point apparatus and are uncorrected. Optical rotations were taken in CHCl_3 using a Jasco DIP-140 polarimeter. Cd measurements were run in dioxane by employing a Jasco J-500 spectropolarimeter at 25° . Uv spectra were measured in EtOH on a Hitachi 150-20 spectrophotometer, and ir spectra were recorded as KBr discs with a Jasco A-100 ir spectrophotometer. ^1H - and ^{13}C -nmr spectra were obtained in CDCl_3 on a Varian XL-300 instrument at 300 MHz and 74.5 MHz, respectively, using TMS as internal standard. Eims were recorded at 70 eV (probe) on a Hitachi M-60 double focusing mass spectrometer. Si gel 60 (Merck, 70–230 mesh) was used for cc. Si gel PF₂₅₄ plates (Merck, 2 mm) were employed for preparative tlc.

ISOLATION OF COMPOUNDS.—Previously, we reported that Si gel cc of the crude crystalline solid (56.5 g) deposited from the syrupy Et₂O extract of the dried stem bark of *A. veitchii* (7.9 kg) led to the isolation of abieslactone [1] and veitchiolide [2] (4). In this stage, resinous products showing closely similar tlc spots have been obtained from the intermediate fractions (fractions 135–165) eluted with CHCl_3 and CHCl_3 -EtOAc (20:1). These were combined and condensed to give an amorphous solid (4.3 g), which was subjected to rechromatography on a Si gel (360 g) column. Elution with CHCl_3 afforded compounds 2 (182 mg) and 3 (53 mg), in order of polarity. Successive cc furnished compounds 6 (1.3 g) and 7 (28 mg) from the fractions eluted with the mixed solvent of CHCl_3 -EtOAc (30:1) and CHCl_3 -EtOAc (20:1), respectively.

COMPOUND 2.—Prisms, mp 245.5 – 248° (EtOAc), $[\alpha]_D^{23} +27.5^\circ$ ($c = 0.55$, CHCl_3); R_f 0.56 [0.25 mm thick, solvent C_6H_6 - CHCl_3 -EtOAc (1:1:1)]; hrms $[\text{M}]^+ 452.3288$ ($\text{C}_{30}\text{H}_{44}\text{O}_3$ requires 452.3290); ir ν max cm^{-1} 3070, 3022, 2930, 2852, 1740 (α, β -unsaturated γ -lactone), 1698 (6-membered ring $\text{C}=\text{O}$), 1648 ($\text{C}=\text{C}$), 1460, 1410 (CH_2CO), 1378, 1360, 1197, 1095, 1040, 933, 878, 818 ($-\text{HC}=\text{C}$); ^1H nmr see Table 1; ^{13}C nmr see Table 2; eims m/z (rel. int.) $[\text{M}]^+ 452$ (46), $[\text{M} - \text{Me}]^+ 437$ (100), $[\text{M} - \text{Me} - \text{H}_2\text{O}]^+ 419$ (13), 315 (11), 314 (12), 313 (11), 299 (20), 271 (18), 243 (20), 139 (9), 97 (37). It was identified by direct comparison (mmp, co-tlc, ir, ^1H -nmr ^{13}C nmr, and eims) with an authentic sample of 3-oxo-9 β -lanosta-7,24-dien-26,23 R -olide isolated previously from the stem bark of *A. firma*.

COMPOUND 3.—Needles, mp 239 – 241° (MeOH), $[\alpha]_D^{23} -58.9^\circ$ ($c = 0.32$, CHCl_3); $R_f = 0.42$ [0.25 mm thick, solvent C_6H_6 - CHCl_3 -EtOAc (1:1:1)]; hrms m/z $[\text{M}]^+ 454.3449$ ($\text{C}_{30}\text{H}_{46}\text{O}_3$ requires 454.3446); ir ν max cm^{-1} (KBr) 3560 (OH), 3065 ($-\text{HC}=\text{C}$), 2948, 2870, 1760 (shoulder), and 1740

(α, β -unsaturated γ -lactone), 1658 (C=C), 1462, 1378, 1362, 1215, 1106, 1058, 1022, 980, 960, 878, 805 (-HC=C<); ^1H nmr see Table 1; ^{13}C nmr see Table 2; eims m/z (rel. int.) $[\text{M}]^+$ 454 (18), $[\text{M} - \text{Me}]^+$ 439 (35), $[\text{M} - \text{H}_2\text{O}]^+$ 436 (4), $[\text{M} - \text{Me} - \text{H}_2\text{O}]^+$ 421 (100), 315 (7), 314 (20), 299 (22), 273 (4), 245 (4), 227 (11), 139 (14), 97 (86). It was identified by direct comparison (mmp, co-tlc, ir, ^1H nmr, ^{13}C nmr, and eims) with an authentic sample of 3 β -hydroxy-9 β -lanosta-7,24-dien-26,23 R -olide isolated also from the bark of *A. firma*.

COMPOUND 6.—Prisms, mp 250–252° (MeOH/CHCl₃), $[\alpha]_D^{25}$ -80.5° ($c = 0.55$, CHCl₃), $R_f = 0.49$ [0.25 mm thick, solvent C₆H₆-CHCl₃-EtOAc (1:1:1)]; hrms m/z $[\text{M}]^+$ 454.3443 (C₃₀H₄₆O₃ requires 454.3446); ir ν max cm⁻¹ (KBr) 3480 (OH), 3048 (-HC=C<), 2932, 2850, 1745, and 1728 (shoulder) (α, β -unsaturated γ -lactone), 1645 (C=C), 1458, 1440, 1375, 1365, 1198, 1087, 1015, 968, 925, 862, 810 (-HC=C<); ^1H nmr see Table 1; ^{13}C nmr see Table 2; eims m/z (rel. int.) $[\text{M}]^+$ 454 (20), $[\text{M} - \text{Me}]^+$ 439 (17), $[\text{M} - \text{H}_2\text{O}]^+$ 436 (5), $[\text{M} - \text{Me} - \text{H}_2\text{O}]^+$ 421 (100), [ion a] 327 (2), [ion b] 315 (11), [ion c] 314 (32), [ion d] 309 (2), [ion e] 299 (20), [ion f] 273 (7), [ion g] 233 (11), 227 (11), [ion h] 189 (3), [ion i] 187 (32), [ion j] 139 (9), [ion k] 109 (30), [ion l] 97 (55); cd (dioxane) $[\theta]_{200} -8100^\circ$, $[\theta]_{208} -26000^\circ$ (trough), $[\theta]_{212} -22200^\circ$, $[\theta]_{220} -11500^\circ$, $[\theta]_{230} -1800^\circ$, $[\theta]_{245} 0^\circ$, $[\theta]_{250} +200^\circ$.

ACETYLATION OF COMPOUND 6.—Compound 6 (100 mg) was dissolved in an equivolume mixture of pyridine and Ac₂O (4 ml). The mixture was kept at room temperature overnight, and subsequent workup afforded 6-acetate (101 mg): m/z $[\text{M}]^+$ 496.3552 (C₃₂H₄₈O₄ requires 496.3552), mp 204.5–208° (MeOH/CHCl₃); ir ν max (KBr) cm⁻¹ 2965, 2948, 2830, 1740, and 1736 (shoulder) (α, β -unsaturated γ -lactone), 1722 (OAc), 1465, 1443, 1382, 1370, 1248 (OAc), 1180, 1091, 1048, 1022, 958, 938, 867, 828; ^1H nmr see Table 1; ^{13}C nmr see Table 2; eims m/z (rel. int.) $[\text{M}]^+$ 496.3552 (14), $[\text{M} - \text{Me}]^+$ 481 (16), $[\text{M} - \text{HOAc}]^+$ 436 (11), $[\text{M} - \text{Me} - \text{HOAc}]^+$ 421 (100), [ion a] 369 (1), [ion b - HOAc and ion f] 315 (7), [ion c] 314 (16), [ion d] 309 (2), [ion e] 299 (14), [ion g] 233 (8), 225 (9), [ion h] 189 (8), [ion i] 187 (20), [ion j] 139 (3), [ion k] 109 (10), [ion l] 97 (2).

OXIDATION OF COMPOUND 6 WITH CrO₃ IN PYRIDINE.—A solution of CrO₃ (50 mg) in pyridine (3 ml) containing 1 drop of H₂O was gradually added to a solution of compound 6 (50 mg) in pyridine (5 ml) at 10°, while stirring, and the mixture was kept at 20° for 5 h. Then, 2 drops of 10% NaHSO₃ solution was added to destroy excess CrO₃. Removal of the pyridine in vacuo gave a residue which was dissolved in Et₂O (30 ml), washed with 5% HCl and H₂O, and dried over Na₂SO₄. Removal of solvent yielded a solid which was purified by preparative tlc [C₆H₆-CHCl₃-EtOAc (1:1:1)] to give the ketone, mp 245–247° (EtOAc), 46.5 mg. The ketone was identified by direct comparison (mmp, co-tlc, ir, ^1H nmr, ^{13}C nmr, and eims) with an authentic sample of 3-oxo-9 β -lanosta-7,24-dien-26,23 R -olide [2] isolated from the stem bark of both *A. firma* and *A. veitchii*.

COMPOUND 7.—Prisms, mp 208–209.5° (MeOH/CHCl₃), $[\alpha]_D^{25}$ +1.4° ($c = 0.65$, CHCl₃); hrms m/z $[\text{M}]^+$ 466.3451 (C₃₁H₄₆O₃ requires 466.3447); uv λ max (EtOH) nm 236, 243.5, 251.5 (log ϵ 4.26, 4.32, 4.15) (heteroannular diene); ir ν max (KBr) cm⁻¹ 2935, 2870, 1758 (shoulder), 1740 (α, β -unsaturated γ -lactone), 1658 (C=C), 1445, 1420, 1377, 1360, 1205, 1100, 1063, 940, 897, 815 (-HC=C<); ^1H nmr see Table 1; ^{13}C nmr see Table 2; eims m/z (rel. int.) $[\text{M}]^+$ 466 (100), $[\text{M} - \text{Me}]^+$ 451 (8), $[\text{M} - \text{MeOH}]^+$ 434 (14), $[\text{M} - \text{Me} - \text{CO}]^+$ 423 (5), $[\text{M} - \text{Me} - \text{MeOH}]^+$ 419 (43), $[\text{M} - \text{Me} - \text{MeOH} - \text{CO}]^+$ 391 (6), 383 (4), 365 (6), 351 (3), [ion m] 339 (6), [ion n] 337 (11), [ion o] 327 (4), [ion p] 325 (5), [ion q] 295 (10), [ion r] 285 (5), [ion s] 253 (19), 239 (11), 227 (9), 225 (12), 213 (8), 211 (8), 201 (13), [ion i] 187 (12), 171 (16), [ion j] 139 (3), [ion k] 109 (8), [ion l] 97 (20); cd (dioxane) $[\theta]_{200} -6700^\circ$, $[\theta]_{208} -19800^\circ$, $[\theta]_{212} -19900^\circ$ (trough), $[\theta]_{230} -7600^\circ$, $[\theta]_{240} -4900^\circ$, $[\theta]_{261} 0^\circ$, $[\theta]_{270} +450^\circ$.

SYNTHESIS OF COMPOUND 7 FROM ABIESLACTONE [1].—A solution of freshly sublimed selenium dioxide (20 mg) in 80% HOAc (5 ml) was gradually added to a solution of abieslactone [1] (40 mg) in HOAc (10 ml), and the mixture was refluxed for 1 h. After cooling, the reaction mixture was diluted with H₂O (40 ml), the resulting precipitate was extracted with Et₂O (50 ml), and the Et₂O layer was neutralized and dried with Na₂SO₄. Removal of the Et₂O afforded a residue (38 mg), which was purified by preparative tlc [C₆H₆-CHCl₃-EtOAc (1:1:1)] to give 3 α -methoxylanosta-7,9(11),24-trien-26,23 R -olide (30.5 mg), mp 206–208.5° (MeOH/CHCl₃), $[\alpha]_D^{25}$ +1.4° ($c = 1.02$, CHCl₃), m/z $[\text{M}]^+$ 466.3445, uv λ max (EtOH) nm 236, 243.5, 251.5 (heteroannular diene). This compound was identified by direct comparison (mmp, co-tlc, uv, ir, ^1H nmr, ^{13}C nmr, and eims) with compound 7.

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